

## Simultaneous Determination of Xipamide and Triamterene by First Derivative, Ratio Difference, and Derivative Ratio Spectrophotometric Methods

Nermine V. Fares<sup>a</sup>, Haitham A. El Fiky<sup>b\*</sup>, Amr M. Badawey<sup>b</sup>, Maha F. Abd El Ghany<sup>a</sup>

<sup>a</sup>Department of Analytical Chemistry, Faculty of Pharmacy, Ain Shams University, Cairo 11566, Egypt

<sup>b</sup>Department of Pharmaceutical Chemistry, Faculty of Pharmaceutical Sciences and Pharmaceutical Industries, Future University in Egypt, Cairo, Egypt

### ABSTRACT

Validated, rapid, and sensitive spectrophotometric techniques were established for the simultaneous determination of Xipamide and Triamterene. The first technique is based on the determination of Triamterene with zero-crossing of Xipamide using the zero-order method at 367.0 nm. The second technique is based on the determination of both Xipamide and Triamterene by the first derivative method with zero-crossing of Triamterene and Xipamide respectively, at 265.6 and 388.6 nm. The third technique is the ratio difference spectrophotometric method depending on obtaining peak amplitude difference at 256.0 and 273.0 nm for Xipamide and 288.0, 302.0 nm for Triamterene. The fourth method is the derivative ratio spectrophotometric method depending on obtaining the first derivative of the ratio spectrum with zero-crossing of Xipamide and Triamterene at 365.2 and 308.6 nm; respectively. Linear relationship was obtained upon using concentration range (1.0-10.0 µg/mL) for Xipamide and (1.0-16.0 µg/mL) for Triamterene with LOD less than 0.3 µg/mL for both drugs. The suggested spectrophotometric techniques showed Lower LOD and more sensitivity other than any reported spectrophotometric methods and were applied in pure and dosage form (Epitens<sup>®</sup>).

**Keywords:** Xipamide; Triamterene; Zero-order; First derivative; Ratio difference; Derivative ratio.

\*Correspondence | Haitham A. El Fiky; Department of Pharmaceutical Chemistry, Faculty of Pharmaceutical Sciences and Pharmaceutical Industries, Future University in Egypt, Cairo, Egypt. Email: [haitham.a.elfiky@gmail.com](mailto:haitham.a.elfiky@gmail.com)

Citation | Fares NV, El Fiky HA, Badawey AM, Abd El Ghany MF, 2021. Simultaneous determination of Xipamide and Triamterene by first derivative, ratio difference, and derivative ratio spectrophotometric methods. Arch Pharm Sci ASU 5(1): 52-62

DOI: [10.21608/aps.2021.58496.1053](https://doi.org/10.21608/aps.2021.58496.1053)

Print ISSN: 2356-8380. Online ISSN: 2356-8399.

Received 25 March 2021. Accepted 26 May 2021.

Copyright: ©2021 Fares *et al.* This is an open-access article licensed under a Creative Commons Attribution 4.0 International License (CC BY 4.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original author(s) and source are credited.

Published by: Ain Shams University, Faculty of Pharmacy

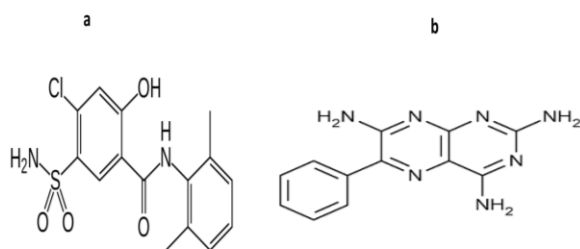
### 1. INTRODUCTION

Xipamide shown in (Fig. 1a) [1] is a (4-chloro-N-(2, 6-dimethyl phenyl)-2-hydroxy-5-sulfamoylbenzamide) (C<sub>15</sub>H<sub>15</sub>Cl N<sub>2</sub>O<sub>4</sub>S) used for the treatment of high blood pressure. Xipamide has an effective action rather than thiazides and decreases the loss of potassium. Triamterene showed in (Fig. 1b) [2, 3] (6-phenylpteridine-2, 4, 7-triamine) (C<sub>12</sub>H<sub>11</sub>N<sub>7</sub>) is a potassium-sparing diuretic preventing hypokalemia in the body and

also used for the treatment of edema. Triamterene showed hyperkalemia as its main side effect. So, to counter this effect it is used in combination with Xipamide to overcome this side effect [4].

The literature review revealed that many methods were reported for determination of Xipamide including Spectrophotometric [5], Spectrofluorometric [6], Voltammetric [7] and HPLC methods [8-10], and for determination of Triamterene including spectrophotometric [11-15], Spectrofluorometric [16-18], Electrochemical

[19, 20], and HPLC methods [21, 22]. Few methods were stated for obtaining both Xipamide and Triamterene in a mixture including Spectrophotometric [23, 24] and HPLC methods [25-27].



**Fig. 1.** Chemical structures of (a) Xipamide and (b) Triamterene

UV-visible spectroscopic methods are quantitative techniques depending on the functional group of the drug. [28]. The functional group of Xipamide (sulfamoyl group) and amino groups of triamterene allow them to absorb Visible UV light within range (200-400 nm). The ratio Difference Spectrophotometric Method [29] depends on the direct proportionality of the amplitude difference and the concentration. The derivative Ratio spectrophotometric method [30] depends on the first derivatization of the ratio spectra and recording at no interference wavelengths.

The proposed methods aim to establish new, selective, rapid, and precise Spectrophotometric techniques to determine Xipamide and Triamterene in pure and dosage forms with lower LOD and higher sensitivity. These methods would ease actual and financial regulation of resources for quality control (QC) aspects.

## 2. Experimental

### 2.1. Apparatus

Double beam Shimadzu UV-Vis spectrophotometer (UV-1800, Japan), Double beam (Libra, biochrom, England), and 1 cm

quartz cells

Sonicator (Branson Model 3510 Ultrasonic Cleaner, UK)

Analytical Balance Sartorius CPA225D, Italy.

## 2.2. Materials and reagents

### 2.2.1. Pure samples

Pure standards of Xipamide and Triamterene were kindly obtained from Epico, Egypt. The purity of Xipamide was tested and found to be  $99.61 \pm 1.206$  according to the reported method [24], while Triamterene was found to be  $100.05 \pm 0.844$  according to the reported method [3].

### 2.2.2. Market sample

Epitens<sup>®</sup> tablet was purchased from the market, manufactured by Epico, Egypt. Epitens<sup>®</sup> was labeled to contain 10.0 mg of Xipamide and 30.0 mg of Triamterene, batch number (11824).

### 2.2.3. Reagents

Methanol (HPLC grade, Sigma, Germany), Methanol (HPLC grade, Merck, Germany), Methanol (HPLC grade, Honeywell, America),

## 2.3. Standard and Working Solutions

### 2.3.1. Xipamide and Triamterene Stock Standard Solution (1000.0 µg/mL)

To prepare stock solutions, 0.1 g of Xipamide and Triamterene were individually dissolved in methanol, transferred to 100-mL volumetric flasks, and diluted with methanol.

### 2.3.2. Xipamide and Triamterene Working Standard Solution (100.0 µg/mL)

10 mL from Xipamide and Triamterene Stock Standard Solution (1000.0 µg/mL) were transferred separately to a 100-mL volumetric flask and diluted with methanol.

## 2.4. Procedures

### 2.4.1. Spectral characteristics of Xipamide and Triamterene

Aliquots of 0.5 mL of Xipamide and Triamterene were individually transferred from their working stock solutions to different 10-mL volumetric flasks and diluted with methanol. The zero-order spectrum of Xipamide and Triamterene (5 µg/mL) was recorded. Triamterene could be determined at 367.0 nm without any interference of Xipamide, while Xipamide could not be determined due to interference of Triamterene.

### 2.4.2. First derivative method

To different 10-mL volumetric flasks, accurate volumes of Xipamide (0.1-1.0 mL) and Triamterene (0.1-1.6 mL) were transferred from their corresponding working standard solutions and diluted with methanol. Obtained concentration ranges were (1.0-10.0 µg/mL) and (1.0-16.0 µg/mL) for Xipamide and Triamterene; respectively. First derivative peak amplitude was recorded at 265.6 nm (zero interference of Triamterene) and 388.6 nm (zero interference of Xipamide), using  $\Delta\lambda = 4$  and scaling factor = 10.

### 2.4.3. Ratio difference method

Different concentrations of Xipamide and Triamterene were prepared within linearity (1.0-10.0 µg/mL) and (1.0-16.0 µg/mL) for Xipamide and Triamterene; respectively. The peak amplitude of Xipamide was recorded without any interference of Triamterene at 256.0 and 273.0 nm after the division of Zero-order spectra of Xipamide by zero-order spectrum of Triamterene (6.0 µg/mL) as Triamterene will be constant so the difference in peak amplitude will equal zero. Repeating these steps for the determination of Triamterene by dividing its spectra by Xipamide (4.0 µg/mL), Triamterene could be determined at peak amplitude 288.0 and 302.0 nm.

### 2.4.4. Derivative ratio method

Different concentrations of Xipamide and Triamterene were prepared within range (1.0-10.0 µg/mL) and (1.0-16.0 µg/mL) for Xipamide and Triamterene; respectively. Spectra of the first derivative were obtained after the division of the Zero-order spectrum of xipamide over Zero-order spectrum of Triamterene (6.0 µg/mL) and Xipamide could be determined at peak amplitude 308.6 nm, while Triamterene could be determined by the same procedures using Xipamide (4.0 µg/mL) as a divisor and recording peak amplitude at 365.2 nm.

### 2.4.5. Laboratory prepared mixtures

Lab mixtures of both drugs were prepared by transferring different volumes within linearity (1.0-10.0 µg/mL) and (1.0-16.0 µg/mL) for Xipamide and Triamterene; respectively from their working standard solutions.

### 2.4.6. Application to a pharmaceutical formulation

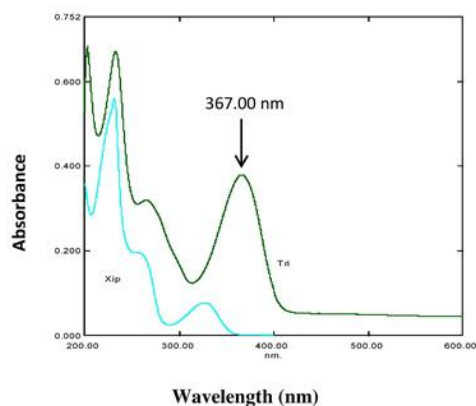
Two tablets of Epitens<sup>®</sup> were ground and mixed. Amounts equal to 10.0 mg of Xipamide and 30.0 mg of Triamterene were accurately transferred to a 100-mL volumetric flask and diluted with methanol. The solution was filtered into the 100-mL volumetric flask to prepare a working standard solution of concentration 100.0 µg/mL of Xipamide and 300 µg/mL of Triamterene. Further dilution was performed to obtain a concentration of 1.0 µg/mL of Xipamide and 3.0 µg/mL of Triamterene.

## 3. Results and Discussion

### 3.1. Spectrophotometric measurement

Studying different solvents and some of the instrumental parameters such as the wavelength increment ( $\Delta\lambda$ ) and the scaling factor were important to review their impact on the shape and resolution of the obtained spectra. DMSO, 0.1 N NaOH, and 0.1 N HCl were tried as solvents for

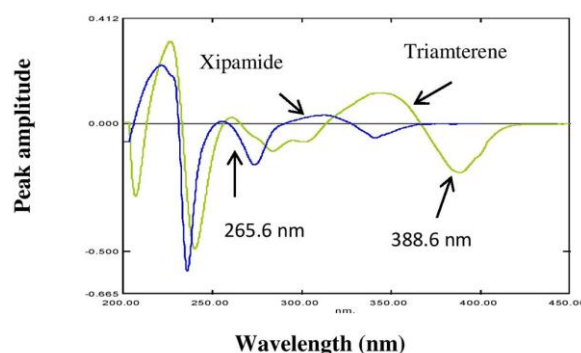
both drugs, different  $\Delta\lambda$  (4, 8, and 16) and scaling factors (10 and 100) were tried, all previously mentioned solvents showed bad resolution and high level of noise. The best spectra for both drugs were obtained upon using methanol as solvent,  $\Delta\lambda = 4$  and scaling factor = 10. After recording the zero-order spectrum of Xipamide and Triamterene, Triamterene can be determined at 367.0 nm without any interference of Xipamide, while Xipamide cannot be determined due to overlapping of Triamterene as revealed in (Fig. 2).



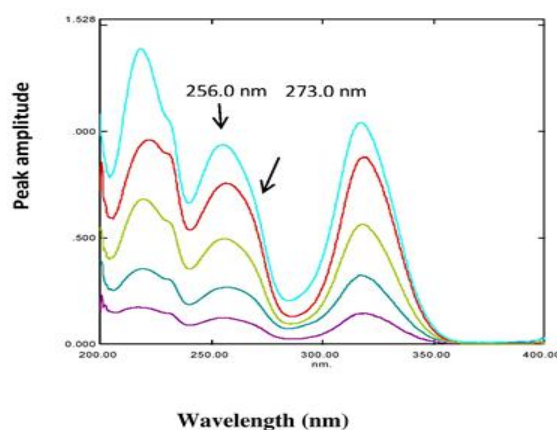
**Fig. 2.** Zero order spectra of both Xipamide and Triamterene (5.0  $\mu\text{g/mL}$  each)

Concerning the first derivative method, the peak amplitude of Xipamide was recorded at 265.6 nm (zero interference of Triamterene) and 388.6 nm for Triamterene (zero interference of Xipamide) as displayed in (Fig. 3). Regarding Ratio difference method, different concentrations of Xipamide (2.0, 4.0, 6.0, 8.0  $\mu\text{g/mL}$ ) and Triamterene (4.0, 6.0, 8.0, 10.0  $\mu\text{g/mL}$ ) were attempted as a divisor, the concentrations showing best resolution and reproducibility were (4.0 and 6.0  $\mu\text{g/mL}$ ) for Xipamide and Triamterene; respectively as presented in (Fig. 4) and (Fig. 5). While for derivative ratio method, peak amplitude for Xipamide and Triamterene was recorded at different wavelengths (234.60, 247.00, 273.40, 308.6 nm) for Xipamide and (279.60, 365.20, 368.00, 384.00 nm) for

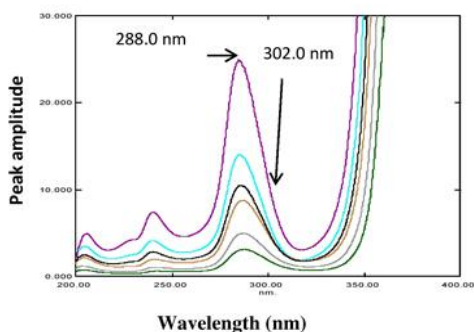
Triamterene with mean recoveries of  $104.61 \pm 2.258$  (234.60 nm),  $102.99 \pm 2.346$  (247.00 nm) and  $103.01 \pm 2.181$  (273.40 nm) for Xipamide, while for triamterene  $103.72 \pm 2.192$  (279.60 nm),  $102.89 \pm 1.913$  (368.00 nm) and finally  $103.60 \pm 2.268$  (384.00 nm), the most sensitive wavelengths with better recovery and least standard deviation were 308.60 ( $100.19 \pm 1.089$ ) and 365.20 ( $100.49 \pm 0.707$ ) for Xipamide and Triamterene; respectively as displayed in (Fig. 6) and (Fig. 7). Fig. 8 and Fig. 9 showed that both Xipamide and Triamterene were superimposed with their synthetic laboratory-prepared mixtures at the selected wavelengths.



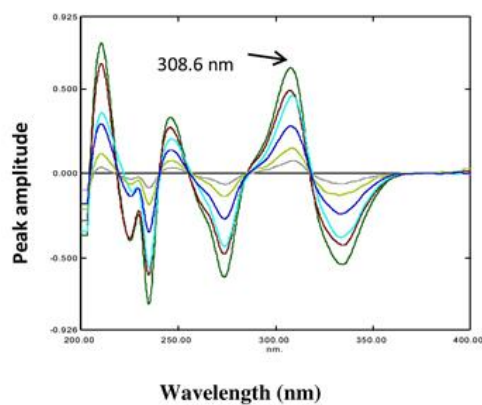
**Fig. 3.** First derivative spectra of different concentrations of both Xipamide and Triamterene at 265.6 & 388.6 nm



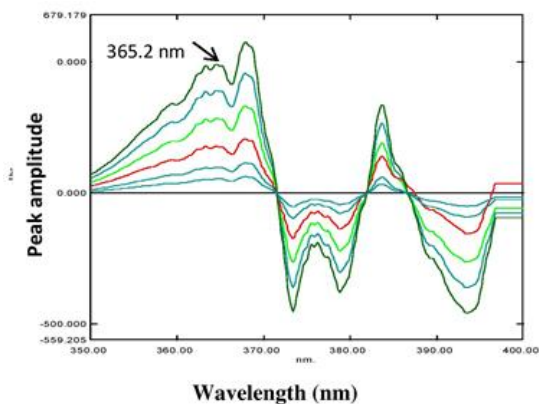
**Fig. 4.** Ratio absorption spectra of intact Xipamide (1.0-10.0  $\mu\text{g/mL}$ ) after resolution from Triamterene (6.0  $\mu\text{g/mL}$ ) using ratio difference method at specified wavelength 256.0 & 273.0 nm



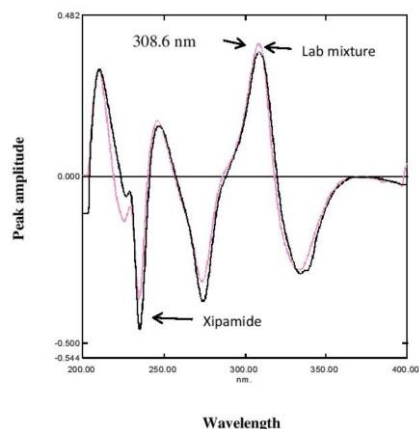
**Fig. 5.** Ratio absorption spectra of intact Triamterene (1.0-16.0  $\mu\text{g/mL}$ ) after resolution from Xipamide (4.0  $\mu\text{g/mL}$ ) using ratio difference method at specified wavelength 288.0 & 302.0 nm



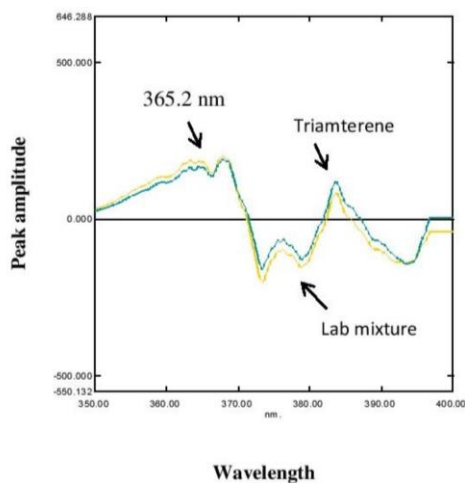
**Fig. 6.** First derivative of ratio spectra for Xipamide (1.0-10.0  $\mu\text{g/mL}$ ) with Triamterene as a divisor (6.0  $\mu\text{g/mL}$ ) at 308.6 nm



**Fig. 7.** First derivative of ratio spectra for Triamterene (1.0-16.0  $\mu\text{g/mL}$ ) with Xipamide as a divisor (4.0  $\mu\text{g/mL}$ ) at 365.2 nm



**Fig. 8.** First derivative of ratio spectra for Xipamide (6.0  $\mu\text{g/mL}$ ) with Laboratory prepared mixture (6.0  $\mu\text{g/mL}$  of Xipamide and 2.0  $\mu\text{g/mL}$  of Triamterene)



**Fig. 9.** First derivative of ratio spectra for Triamterene (4.0  $\mu\text{g/mL}$ ) with Laboratory prepared mixture (8.0  $\mu\text{g/mL}$  of Xipamide and 4.0  $\mu\text{g/mL}$  of Triamterene)

## 3.2. Method Validation

### 3.2.1. Linearity

Xipamide and Triamterene showed direct relation between concentrations and peak amplitude within range (1.0-10.0  $\mu\text{g/mL}$ ) and (1.0-16.0  $\mu\text{g/mL}$ ); respectively.

### 3.2.2. Accuracy

Xipamide and Triamterene concentrations were determined using the corresponding regression equation. The obtained results showed

good accuracy with mean recovery less than 100.90 and standard deviation less than 1.53 for both drugs as displayed in **Table 1**.

**Table 1. Validation parameters for Spectrophotometric determination of Xipamide and Triamterene**

Parameter	Xipamide				Triamterene		
	First derivative	Ratio difference	Derivative ratio	Zero order	First derivative	Ratio difference	Derivative ratio
<b>Linearity :</b>							
Range (µg/mL)	1.0-10.0 µg/mL				1.0-16.0 µg/mL		
Intercept	0.0005	0.0037	0.0135	0.0597	0.0011	0.882	19.687
Slope	0.0128	0.0521	0.0654	0.0758	0.0305	0.7861	36.077
Correlation coefficient(r)	0.9996	0.9999	0.9997	0.9999	0.9997	0.9999	0.9998
SD of the residuals	0.000935	0.00191	0.003623	0.003679	0.002829	0.049757	3.094408
SE of intercept	0.000813	0.001663	0.003152	0.002702	0.002078	0.036545	2.27275
Slope (X coefficient)	0.012774	0.052096	0.065361	0.075808	0.030532	0.786072	36.07661
SE of slope (X coefficient)	0.000134	0.000274	0.000519	0.000326	0.000250	0.004405	0.273938
Accuracy (mean±SD)	99.55±0.907	99.0±0.820	100.19±1.089	99.43±0.939	100.82±0.473	99.90±1.529	100.49±0.707
<b>Precision:</b>							
Repeatability*	0.813	1.373	1.179	0.884	1.034	0.900	1.045
Intermediate precision**	1.272	1.506	1.263	1.028	1.136	1.274	1.194
LOD (µg/mL)***	0.241	0.121	0.183	0.160	0.306	0.209	0.283
LOQ (µg/mL)****	0.730	0.367	0.554	0.485	0.928	0.633	0.858

\*The intra-day precision (n= 3), average of three different concentrations (2.0, 4.0 and 6.0 µg/mL) and (4.0, 6.0 and 10.0 µg/mL) for both Xipamide and Triamterene; respectively repeated three times within day

\*\*The inter-day precision ( = 3), average of three different concentrations (2.0, 4.0 and 6.0 µg/mL) and (4.0, 6.0 and 10.0 µg/mL) for both Xipamide and Triamterene; respectively repeated three times in three successive days

\*\*\*Limit of detection and\*\*\*\* limit of quantitation are determined via calculations

LOD= (SD of the response/slope) × 3.3.

LOQ= (SD of the response/slope) × 10

### 3.2.3. Precision

On three successive days and within a day, different concentrations of Xipamide (2.0, 4.0, and 6.0 µg/mL) and Triamterene (4.0, 6.0, and 10.0 µg/mL) were analyzed to check Precision as illustrated in **Table 1**.

### 3.2.4. Specificity

Specificity was done by using different concentration ratios of Xipamide and Triamterene within linearity. The mean recovery results of Xipamide and Triamterene by the suggested spectrophotometric techniques were presented in **Table 2** showing excellent selectivity.

**Table 2. Determination of Xipamide and Triamterene in the laboratory prepared mixtures by Spectrophotometric methods**

Binary mixture Xipamide: Triamterene ratios	First derivative %Recovery*		Ratio difference %Recovery*		Derivative ratio %Recovery*		Zero order %Recovery*
	Xipamide	Triamterene	Xipamide	Triamterene	Xipamide	Triamterene	Triamterene
3: 1	98.75	99.75	101.75	99.92	98.75	99.42	98.42
2: 1	99.67	101.42	99.67	98.25	100.00	100.33	100.42
1: 1	98.56	100.56	99.00	98.89	101.89	101.33	99.11
1: 2	98.13	98.50	101.75	98.75	100.50	98.50	101.00
1: 3	100.83	100.00	99.67	99.50	98.17	100.00	98.50
<b>Mean ± SD</b>	<b>99.19±1.076</b>	<b>100.05±1.076</b>	<b>100.37±1.291</b>	<b>99.06±0.655</b>	<b>99.86±1.470</b>	<b>99.92±1.052</b>	<b>99.49±1.163</b>

\*Average of three determinations

**Table 3. Statistical comparison of the results of the proposed Spectrophotometric methods, the Reported and the Official methods for determination of Xipamide and Triamterene**

Value	First derivative	Ratio difference	Derivative ratio	Reported method*	Zero order	First derivative	Ratio difference	Derivative ratio	Official method**
	Xipamide	Xipamide	Xipamide	Xipamide	Triamterene	Triamterene	Triamterene	Triamterene	Triamterene
Mean	99.55	99.00	100.19	99.61	99.43	100.82	99.90	100.49	100.05
SD	0.907	0.820	1.089	1.206	0.939	0.473	1.529	0.707	0.844
N	5	5	5	4	5	5	5	5	4
V (variance)	0.823	0.672	1.186	1.454	0.882	0.224	2.338	0.500	0.712
Student's-t test*** (2.365)	0.083	0.864	0.748		1.041	1.631	0.187	0.834	
F-test***	1.768 (6.590)	2.163 (6.590)	1.226 (6.590)		1.239 (9.12 0)	3.184 (6.590)	3.282 (9.120)	1.425 (6.590)	

\*Ratio subtraction spectrophotometric method [24] for determination of Xipamide

\*\*BP determination of Triamterene by Potentiometric method [3]

\*\*\*The values in the parenthesis are the corresponding theoretical values of t and F at (p=0.05). No significant difference by using one way ANOVA with F equals 1.176 ( $F_{crit}=3.287$ ), F equals 1.518 ( $F_{crit}=2.895$ ) and p equals (0.352, 0.237) for Xipamide and Triamterene; respectively

### 3.2.5. Statistical analysis

The proposed spectrophotometric methods were statistically compared to the Ratio subtraction spectrophotometric technique for Xipamide [24] and the Official Potentiometric method for Triamterene [3] according to ICH guidelines [31, 32] as illustrated in Tables 3 with insignificant changes in the produced results.

### 3.2.6. Robustness

The robustness of the proposed methods was evaluated by studying different parameters including different devices (Libra, biochrom, England), different lots of solvent (Merck, sigma, and Honeywell), and different wavelengths. The obtained results were acceptable and displayed in Table 4

**Table 4. Robustness results of the proposed methods**

Parameters	First derivative		Ratio difference		Derivative ratio	
	Xipamide	Triamterene	Xipamide	Triamterene	Xipamide	Triamterene
	Recovery $\pm$ RSD		Recovery $\pm$ RSD		Recovery $\pm$ RSD	
Different device	101.35 $\pm$ 0.541	100.60 $\pm$ 0.384	99.05 $\pm$ 0.620	100.57 $\pm$ 0.321	98.92 $\pm$ 0.672	99.93 $\pm$ 0.462
Lots of solvent	101.13 $\pm$ 0.372	100.97 $\pm$ 0.282	100.12 $\pm$ 0.623	100.74 $\pm$ 0.698	98.92 $\pm$ 0.736	100.08 $\pm$ 0.377
Wavelength*	100.91 $\pm$ 1.165	100.51 $\pm$ 0.411	99.58 $\pm$ 1.099	100.81 $\pm$ 0.238	99.35 $\pm$ 0.922	100.39 $\pm$ 0.773

\*peak amplitude was recorded at the selected wavelengths  $\pm$  1.0 for the proposed methods except for first derivative method for determination of Xipamide at  $\pm$ 0.2

**Table 5. Analysis of pharmaceutical dosage form (Epitens<sup>®</sup>) and application of standard addition technique by the proposed Spectrophotometric methods for determination of Xipamide**

Epitens <sup>®</sup> Recovery* % $\pm$ SD	First derivative			Epitens <sup>®</sup> Recovery* % $\pm$ SD	Ratio difference			Epitens <sup>®</sup> Recovery* % $\pm$ SD	Derivative ratio		
	Standard addition				Standard addition				Standard addition		
	Pure taken $\mu$ g/mL	Pure found $\mu$ g/mL	Recovery %*		Pure taken $\mu$ g/mL	Pure found $\mu$ g/mL	Recovery %*		Pure taken $\mu$ g/mL	Pure found $\mu$ g/mL	Recovery %*
98.33 $\pm$ 0.577	0.50	0.49	98.00		0.50	0.50	100.00		0.50	0.50	100.00
	1.00	1.01	101.00	99.33 $\pm$	1.00	0.99	99.00	99.33 $\pm$	1.00	1.01	101.00
	2.00	2.03	101.50	1.155	2.00	1.98	99.00	0.577	2.00	2.00	100.00
	Mean $\pm$ SD		100.17 $\pm$ 1.893		Mean $\pm$ SD		99.33 $\pm$ 0.577		Mean $\pm$ SD		100.33 $\pm$ 0.577

\*Average of three determinations

### 3.3. Application to a pharmaceutical formulation

Xipamide and Triamterene were successfully determined in pharmaceutical form (Epitens<sup>®</sup>) using the suggested spectrophotometric techniques without any interference from the excipients in tablets. There was no need for pretreatment or any extraction of the sample. The obtained mean recoveries were acceptable for

Xipamide and Triamterene. Standard addition technique was applied to validate the developed methods as displayed in Tables 5 and Table 6. Also, a comparison between the proposed and the reported methods were shown in Table 7.

### Conclusion

UV spectrophotometric methods play an important role in the determination and analysis of many components in analytical fields due to



many advantages such as accuracy of the device, is simple, ease of operation, and low solvent consumption. Determination of Xipamide and

Triamterene by the proposed spectrophotometric methods showed high selectivity in pure forms and pharmaceutical form.

**Table 6. Analysis of pharmaceutical dosage form (Epitens®) and application of standard addition technique by the proposed Spectrophotometric methods for determination of Triamterene**

Epitens® Recovery * % ±SD	Zero order			Epitens® Recovery * % ±SD	First derivative			Epitens® Recovery * % ±SD	Ratio difference			Epitens® Recovery * % ± SD	Derivative ratio		
	Standard addition				Standard addition				Standard addition				Standard addition		
	Pure taken µg/mL	Pure found µg/mL	Recovery %*		Pure taken µg/mL	Pure found µg/mL	Recovery %*		Pure taken µg/mL	Pure found µg/mL	Recovery %*		Pure taken µg/mL	Pure found µg/mL	Recovery %*
96.67± 1.858	1.50	1.51	100.67		1.50	1.51	100.67		1.50	1.53	102.00		1.50	1.52	101.33
	3.00	2.98	99.33	100.67 ± 1.000	3.00	2.98	99.33	96.22±1. 169	3.00	3.05	101.67	99.33±1. 732	3.00	2.95	98.33
	6.00	6.02	100.33		6.00	5.97	99.50		6.00	6.12	102.00		6.00	5.91	98.50
	Mean±SD		100.11± 0.697		Mean±SD		99.83 ±0.730		Mean±SD		101.89± 0.191		Mean±SD		99.39± 1.685

\*Average of three determinations

**Table 7. A comparison between the proposed and the reported methods**

Parameters	Xipamide				Triamterene					
	First derivative	Ratio difference	Derivative ratio	Reported spectro	Reported HPLC	First derivative	Ratio difference	Derivative ratio	Reported spectro	Reported HPLC
LOD	0.241	0.121	0.183	0.76	0.09	0.306	0.209	0.283	0.81	0.06
selectivity	99.19± 1.076	100.37± 1.291	99.86± 1.470	100.3± 1.29	100.75± 0.69	100.05± 1.076	99.06± 0.655	99.92± 1.052	100.3± 0.75	100.92± 0.44
Linearity range µg/mL <sup>-1</sup>	1-10		2-10	0.1-20	1-16		2-12	0.2-50		
Cost and greenness	Low cost and green as methanol is more green than acetonitrile				High cost	Low cost and green as methanol is more green than acetonitrile				High cost

## Declarations

### Ethics approval and consent to participate

Not applicable

### Consent to publish

Not applicable

### Availability of data and materials

The data generated or analyzed during this study all are included in the main manuscript.

## Competing interests

The authors declare that they have no competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

## Funding statement

No funding source was received

## 4. REFERENCES

1. Hempelmann FW. Studies on Xipamide (4-chloro-5-sulfamoyl-2', 6'-salicyloxylydide). Part 1:

- Physico-chemical and chemical properties (author's transl). *Arzneimitt. Forsch.* 1977; 27(11): 2140. <https://doi.org/10.1007/s10895-013-1301-z>
- Brittain HG. Analytical profiles of drug substances and excipients: Academic Press; 1994.
  - Pharmacopoeia B. British pharmacopoeia commission . London, TSO, UK., 1105, 1106; for Triamterene, A-290. 2016: 4.
  - Joel GH, Perry BM, Lee EL, Raymond WR. Alfred Cg, editor. Goodman Gilman's The pharmacological Basis of therapeutics. 9th ed. New Jersey. Mc-Graw Hill Companies; 1996.
  - Gaber M, Khedr AM, El-Kady AS. New and sensitive spectrophotometric method for determination of xipamide in pure and dosage forms by complexation with Fe (III), Cu (II), La (III), UO<sub>2</sub> (II), Th (IV), and ZrO (II) ions. *Int Res J Pharm Pharmacol.* 2011; 1(9): 215-20.
  - Walash MI, El-Enany N, Eid MI, Fathy ME. Stability-indicating spectrofluorimetric methods for the determination of metolazone and xipamide in their tablets. Application to content uniformity testing. *J. Fluoresc.* 2014; 24(2): 363-76. <https://doi.org/10.1007/s10895-013-1301-z>
  - Attia AK, Hendawy HM. Electrochemical Characterization of Xipamide Using Cyclic and Square Wave Voltammetry. *Res Rev Electrochem.* 2017; 8(1):104.
  - Rajendran R, Devikasubramaniyan, Sura RS, Kumar AA, Mounika A, Sowjanya A, et al. Estimation of Xipamide by using HPLC in the pure and pharmaceutical dosage form. *Indo Am. j. pharm.* 2018; 5(4): 2095-102. <http://doi.org/10.5281/zenodo.1214540>
  - Tolba MM, Belal F. Two liquid chromatographic approaches for the simultaneous determination of xipamide and its degradation product (2, 6-xylidine) using time-programmed fluorescence detection. *Luminescence.* 2017; 32(4): 491-501. . <https://doi.org/10.1002/bio.3203>
  - Legorburu M, Alonso R, Jimenez RJJolc, technologies r. Determination of the non-thiazide diuretic xipamide in pharmaceuticals and urine by HPLC with amperometric detection. *J. Liq. Chromatogr. Relat.* 1999; 22(1): 735-46. <https://doi.org/10.1081/JLC-100101695>
  - Hu Y, Wu H-L, Yin X-L, Gu H-W, Kang C, Xiang S-X, et al. Chemometrics-assisted determination of amiloride and triamterene in biological fluids with overlapped peaks and unknown interferences. *Bioanalysis.* 2015; 7(13): 1685-97. <https://doi.org/10.4155/bio.15.88>
  - Stolarczyk M, Apolo A, Krzek J, Lech K. Simultaneous determination of triamterene and hydrochlorothiazide in tablets using derivative spectrophotometry. *Acta Pol. Pharm.* 2008; 65(3). <https://doi.org/10.1111/j.2042-7158.1998.tb00333.x>
  - El Ragehy NA, Abbas SS, El-Khateeb SZ. Spectrophotometric determination of triamterene using some acid dyes. *Polym Plast Technol Eng.* 1995; 28(10): 1799-809. <https://doi.org/10.1080/00032719508000358>
  - El Ragehy NA, Abbas SS, El-Khateeb SZ. Utility of p-chloranilic acid and 2, 3 dichloro-5, 6-dicyano p-benzoquinone (DDQ) for the spectrophotometric determination of triamterene. *Anal. Lett.* 1997; 30(11): 2045-58. <https://doi.org/10.1080/00032719708001720>
  - Kargosha K, Sarrafi AHM. Spectrophotometric simultaneous determination of triamterene and hydrochlorothiazide in Triamterene-H tablets by multivariate calibration methods. *J. Pharm. Biomed. Anal.* 2001; 26(2): 273-9. [https://doi.org/10.1016/S0731-7085\(01\)00412-5](https://doi.org/10.1016/S0731-7085(01)00412-5)
  - Domínguez-Vidal A, Ortega-Barrales P, Molina-Díaz A. Determination of triamterene by transitory retention in a continuous flow solid-phase system with fluorimetric transduction. *J. Pharm. Biomed. Anal.* 2002; 28(3-4): 721-8. [https://doi.org/10.1016/S0731-7085\(01\)00682-3](https://doi.org/10.1016/S0731-7085(01)00682-3)
  - Tabrizi AB, Naini S, Parnian K, Mohammadi S, Anvarian SP, Abdollahi A. Determination of triamterene in human plasma and urine after its cloud point extraction. *Quím. Nova.* 2014; 37(7): 1182-7. <http://dx.doi.org/10.5935/0100-4042.20140188>

18. Pulgarín JAM, Molina AA, López PF. Direct analysis of amiloride and triamterene mixtures by fluorescence spectrometry using partial-least squares calibration. *Anal. Chim. Acta.* 2001; 449(1-2): 179-87. [https://doi.org/10.1016/S0003-2670\(01\)01356-3](https://doi.org/10.1016/S0003-2670(01)01356-3)
19. Hudari FF, Souza JC, Zanoni MVB. Adsorptive stripping voltammetry for simultaneous determination of hydrochlorothiazide and triamterene in hemodialysis samples using a multi-walled carbon nanotube-modified glassy carbon electrode. *Talanta.* 2018; 179: 652-7. <https://doi.org/10.1016/j.talanta.2017.11.071>
20. Nezhadali A, Mojarrab M. Fabrication of an electrochemical molecularly imprinted polymer triamterene sensor based on multivariate optimization using multi-walled carbon nanotubes. *J. Electroanal. Chem.* 2015; 744: 85-94.
21. Jain R, Majithia S, Mayur P, Chaudhary R, Pareek K. RP-HPLC method development and validation for simultaneous estimation of Benzthiazide and Triamterene in their combined dosage form. *World J. Pharm. Res.* 2017. <https://doi.org/10.1016/j.jelechem.2015.03.010>
22. Li H, He J, Liu Q, Huo Z, Liang S, Liang Y. Simultaneous analysis of hydrochlorothiazide, triamterene, and reserpine in rat plasma by high-performance liquid chromatography and tandem solid-phase extraction. *J. Sep. Sci.* 2011; 34(5): 542-7. <https://doi.org/10.1002/jssc.201000754>
23. Nassar MWI, Attia KAM, Mohamad AA, Said RAM, Abdel-Monem AH. Simultaneous spectrophotometric estimation of triamterene and xipamide in pure form and pharmaceutical formulation by iso-absorptive point-dependent methods. *Anal. Chem. Lett.* 2017; 7(6): 792-804. <https://doi.org/10.1080/22297928.2017.1395293>
24. Wagieh NE, Abbas SS, Abdelkawy M, Abdelrahman MM. Spectrophotometric and spectrodensitometric determination of triamterene and xipamide in pure form and pharmaceutical formulation. *Drug Test Anal.* 2010; 2(3): 113-21. <https://doi.org/10.1002/dta.92>
25. Abd El-Hay SS, Hashem H, Gouda AA. High-performance liquid chromatography for simultaneous determination of xipamide, triamterene, and hydrochlorothiazide in bulk drug samples and dosage forms. *Acta Pharm. Sin. B.* 2016; 66(1): 109-18. <https://doi.org/10.1515/acph-2016-0022>
26. El-Kimary EI. Stability-indicating HPLC-DAD method development, validation, and stress degradation studies for triamterene and xipamide in their combined tablet dosage form. *Acta Chromatogr.* 2016; 28(1): 79-98. <https://doi.org/10.1556/achrom.28.2016.1.7>
27. Maher HM, Youssef RM, Eman I, Hassan EM, Barary MA. Bioavailability study of triamterene and xipamide using urinary pharmacokinetic data following a single oral dose of each drug or their combination. *J. Pharm. Biomed. Anal.* 2012; 61: 78-85. <https://doi.org/10.1016/j.jpba.2011.11.032>
28. Gorog. S: Ultraviolet-visible spectrophotometry in pharmaceutical analysis: CRC press. 2018. <https://doi.org/10.1201/9781351077422>
29. Attia, K.A., El-Abasawi, N.M., El-Olemy, A. and Serag, A. Different spectrophotometric methods applied for the analysis of simeprevir in the presence of its oxidative degradation product: A comparative study. *Spectrochim. Acta A Mol. Biomol. Spectrosc.* 2018; 190: 1-9. <https://doi.org/10.1016/j.saa.2017.08.066>
30. Mohamed, A.A., El-Olemy, A., Ramzy, S., Abdelazim, A.H., Omar, M.K. and Shahin, M. Spectrophotometric determination of lesinurad and allopurinol in recently approved FDA pharmaceutical preparation. *Spectrochim. Acta A Mol. Biomol. Spectrosc.* 2021; 247: 119106. <https://doi.org/10.1016/j.saa.2020.119106>
31. Mendham J. *Vogels textbook of quantitative chemical analysis*: Pearson Education India; 2006.
32. Remington JP. *Remington: The science and practice of pharmacy*: Lippincott Williams & Wilkins; 2006.